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CONCEPTS IN..

Dissolving:

Dissociation,

Dispersion or

Ionization Substances

can dissolve in water

by three ways: 1.

dissociation 2.

dispersion 3.

ionization Dissociation

is the process by

which an ionic

compound separates

into ions Ex: NaCl

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Dispersion is the process by which a compound breaks into small pieces that spread throughout the water Ex: sugar dissolves in water
Ionization is the process by which neutral molecules gain or lose electrons (forming ions) Ex: hydrogen chloride gas & water react to ...

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Chapter 4 Atomic

Structure Atom V

Element ATOMIC

THEORY Democritus

(4th century B.C) said

the universe was

made of invisible units

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called atoms In 1808,
an Englishman
named John Dalton
proposed a theory: 1.
every element is
made of tiny unique
particles called atoms
that cannot be
subdivided 2. atoms
of the same element
have the same ...

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Solids have a definite
shape and volume
because particles in a
solid vibrate about a
fixed point

Adding/taking Energy
out of a system When

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you add energy to a system the particles begin to move faster As they move faster they collide and create friction (Heat) As they continue we get a phase change Six Common Phase Changes phase change: is the reversible physical change that occurs when a substance

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changes from one
state of matter to
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phase changes:

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