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STUDYBLUE 5. The electrode at which oxidation occurs is called the a) anode b) cathode c) salt bridge 6. The half-

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(25.2/100) / 0.600 = 0.420 M 2. B 3. C
K₂SO₄ has three ions 2K⁺ and SO₄²⁻ and lowers the freezing point more.
4. D 5. C 6. B 7. A $\ln(0.09) = -9.2 \times 10^{-3} t + \ln(0.1)$ solve for t = 12 sec 8. B
reaction will shift to left, decreases the amounts of CO₂ and CaO 9.

Houston Community College System
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